CFD simulation of water curtain for gas dispersion

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Outline

- Current situation of the Japanese chemical industry
- Conventional methods of gas diffusion analytical
- Gas diffusion analysis using numerical method
 The effects of gas density and wind speed
 The effects of mitigation equipment
- Summary



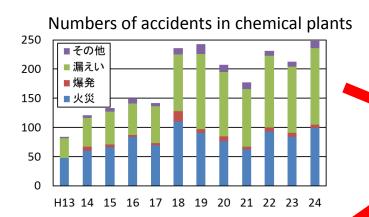
Current situation of the Japanese Chemical industry



Situation of accident in chemical plant in Japan

■In recent years, disasters and accidents occur frequently

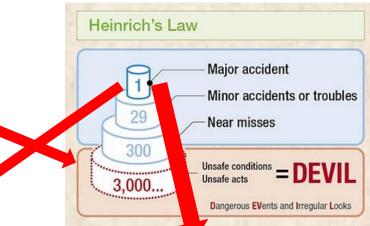
The main cause of the damage expansion is spreading and diffusion of stored items



Flixborough disaster(Cyclohexene)

28 deaths 36 serious injuries





Bhopal Disaster



500,000+ injuries 8000+ deaths

Large leakage accidents make a serious impact on the surrounding area due to toxic or explosive characteristic of storaged substances

Analysis of the leaked gas diffusion

Close distances from factory areas to residential areas





Precise analysis of leaked gas diffusion is required

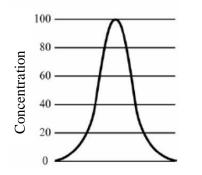


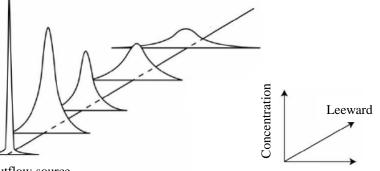
Analytical methods of gas diffusion analytical



Examples of conventional diffusion analytical methods

Density of leaked gas is analogized with Gaussian distribution



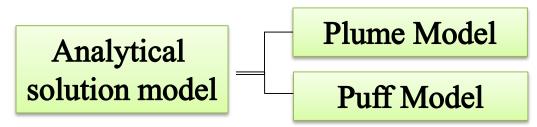


Distance direction perpendicular to the wind

Outflow source

Direction perpendicular to the wind

The 2 of most commonly used analytical methods

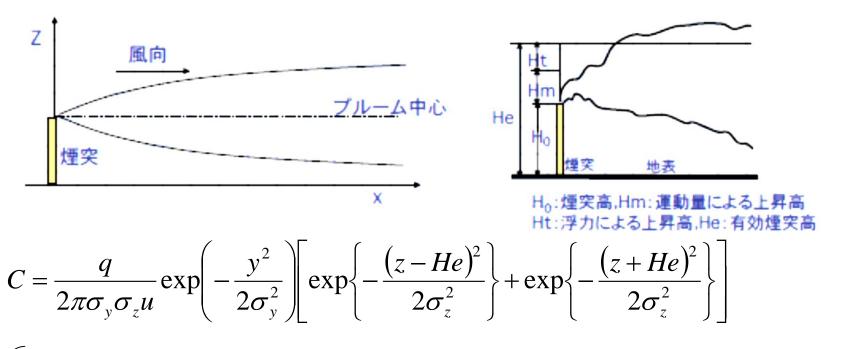




Analytical method based on the Gaussian distribution

Plume Model

→Applied when neutral gas is continuously discharged in a sufficiently large space, and diffuses under one direction with constant velocity.



C=gas concentration (kg/m³) σ y:horizontal standard deviation of the emission distribution(m) q=gas source emission rate (kg/s) σ z:vertical standard deviation of the emission distribution(m) u=horizontal wind velocity along the plume centerline(m/s) He=height of emission plume centerline above ground level(m) x,y,z = measuring point

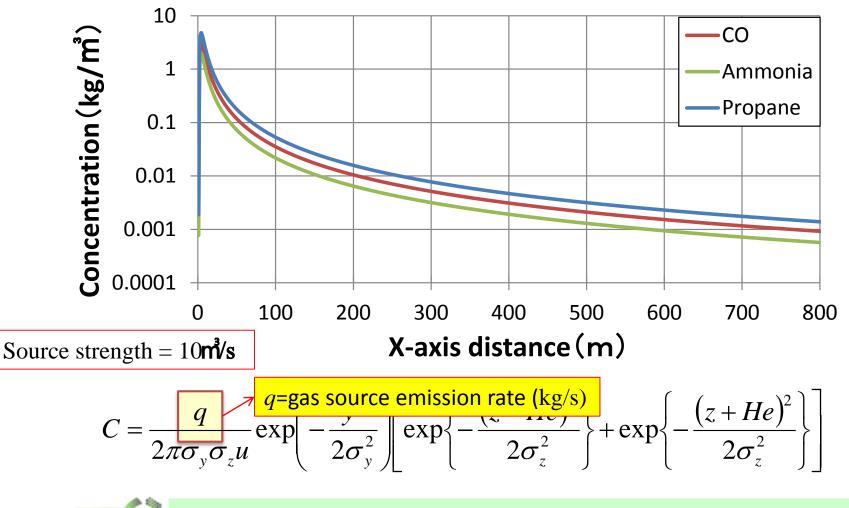
Study gases and analysis conditions

Gas	Gas Density	Source Strength		Wind Spe	eed	
Neutral Gas(CO)	1.25 kg /m ³	10 m³/s	1m/s	3m/s	6m/s	
Lift Gas (Ammonia)	0.771 kg /m³	10 m³/s	1m/s	3m/s	6m/s	3 cases of
Heavy Gas(Propane)	1.882 kg /m ³	10 m³/s	1m/s	3m/s	6m/s	wind speed
3 different density gas		Air de	ensity			

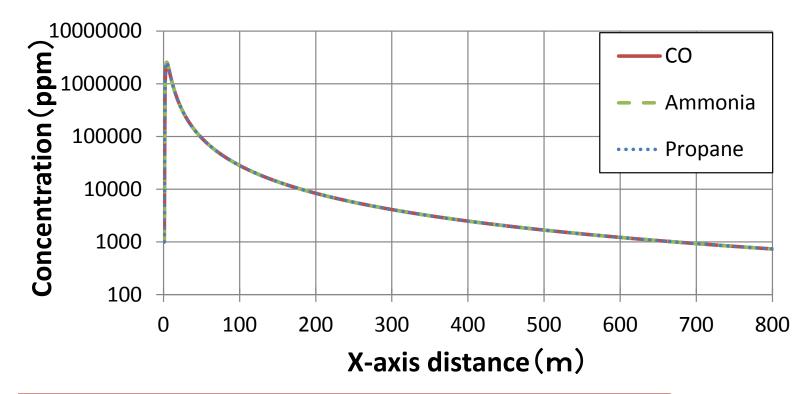
Height of gas source from ground = 0.5m



Gas concentration at ground along x axis(leeward direction) at wind speed=3m/s



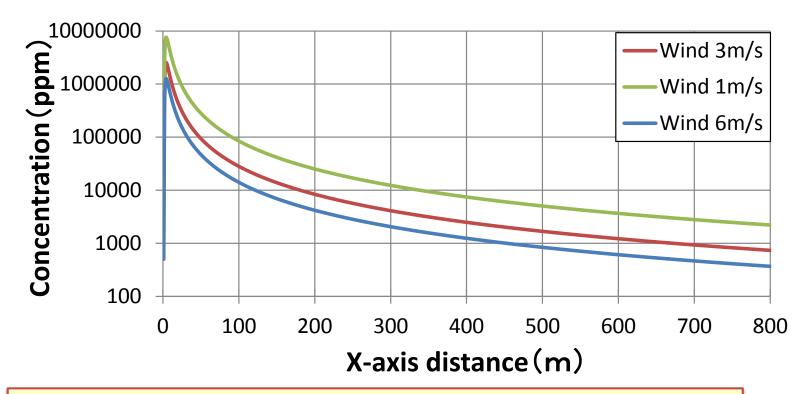
Gas concentration at ground along x axis(leeward direction) at wind speed=3m/s



Plume model does not considers the effects of density



Gas concentration at ground along x axis(leeward direction) at wind speed=1m/s, 3m/s, 6m/s



Gas concentration only changes with wind speed by ratio
 ⇒ Plume model can not considers the effects of wind speed on gas diffusion behavior



Limit of plume model

- -Does not considers
- the effects of density
- the effects of wind speed on gas diffusion behavior
- -Can not evaluate properly at near the gas source
- -Not be able to considers the effects of mitigation equipment

 \Rightarrow We employed numerical method for this research



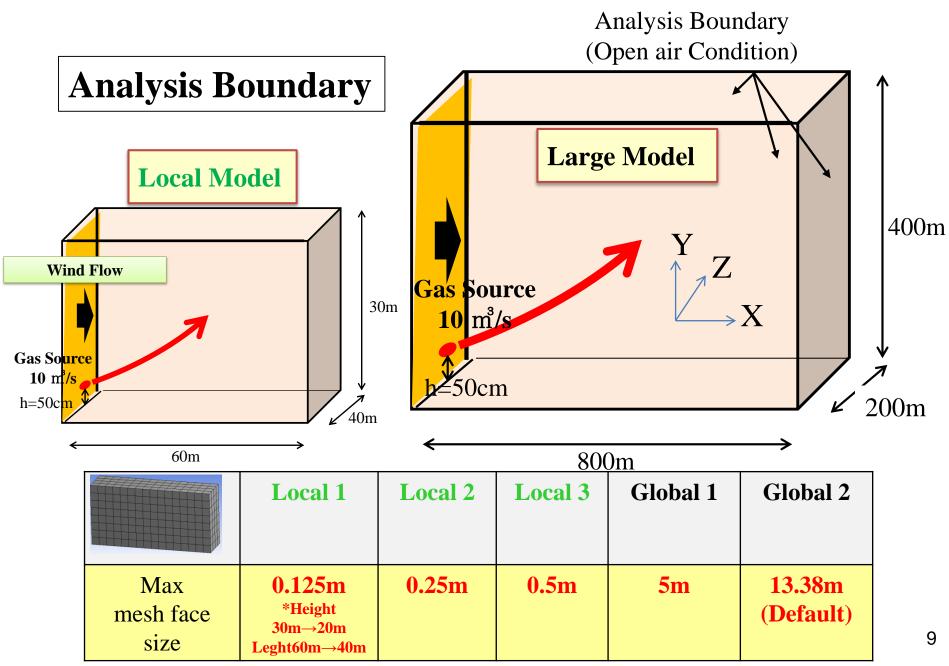
Gas diffusion analysis using numerical method



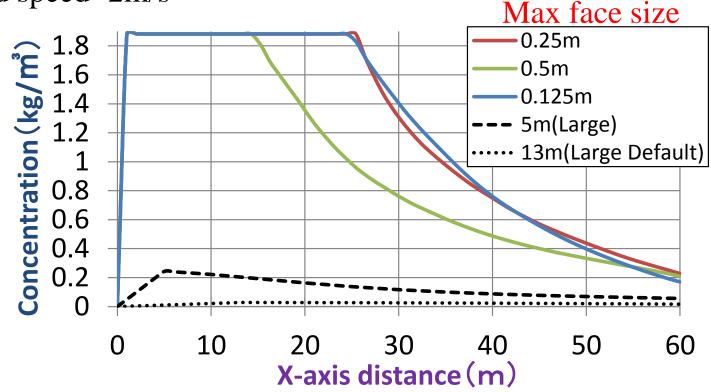
Gas diffusion analysis using numerical method

• CFD simulation software: ANSYS-CFX R16





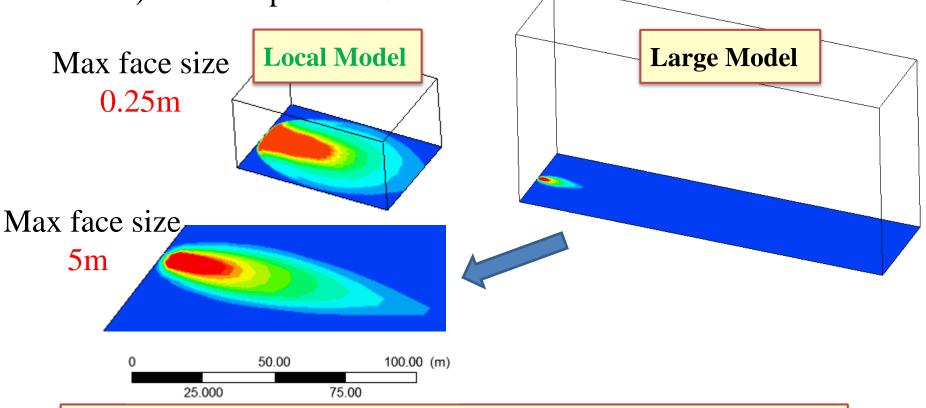
Propane gas concentration at ground along x axis(leeward direction) at wind speed=2m/s



Mesh size has large effect to the gas concentration ⇒To estimate gas concentration in wide area , It is a future challenge to adapt local model with large model.



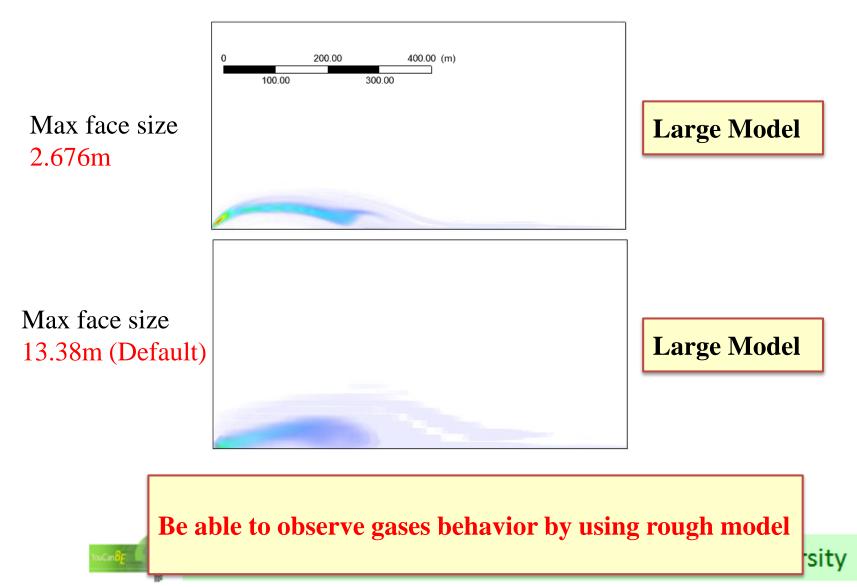
Propane gas distribution at ground along x axis(leeward direction) at wind speed=2m/s



Even it is different mesh size , it is still be able to get similar result of gas dispersion



Ammonia gas dispersion at wind speed=3m/s



Study gases and analysis conditions



Purposes of study

1)Study the effects of density and 2)Simulate leaked gas dispersion with wind speed on gas diffusion behavior considering the effect of water spray curtain

	Gas	Gas Density	Source Strength	Wind Speed		Absorption Level	
	Neutral Gas(CO)	1.25 kg /m ³	10 m³/s	1m/s	3m/s	6m/s	No Curtain
D -	Lift Gas (Ammonia)	0.771 kg /m ^³	10 m³/s	1m/s	3m/s	6m/s	No Curtain
	Heavy Gas(Propane)	1.882 kg /m ³	10 m³/s	1m/s	3m/s	6m/s	No Curtain
	Supposed Gas A	0.771kg/m³	10 m³/s	3m/s			No Curtain
2-	Supposed Gas B	0.771kg/m³	10 m³/s	3m/s		None	
	Supposed Gas C	0.771kg/m³	10 m ³ /s		3m/s		High

Gas diffusion analysis using numerical method

① The effects of wind speed and gas density



(1)Study gases and analysis conditions

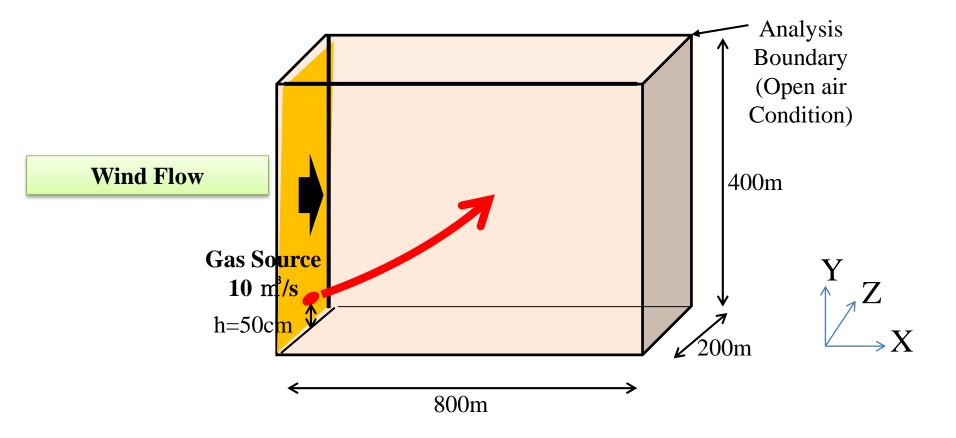
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Heavy Gas(Propane)	1.882 kg/m ^³	10 m³/s	1m/s	3m/s	6m/s	wind speed
3 different	t density gas					

Air density = 1.18 kg/m^3



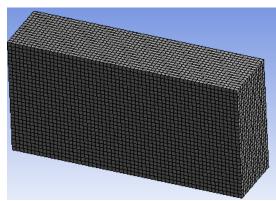
Analysis Boundary



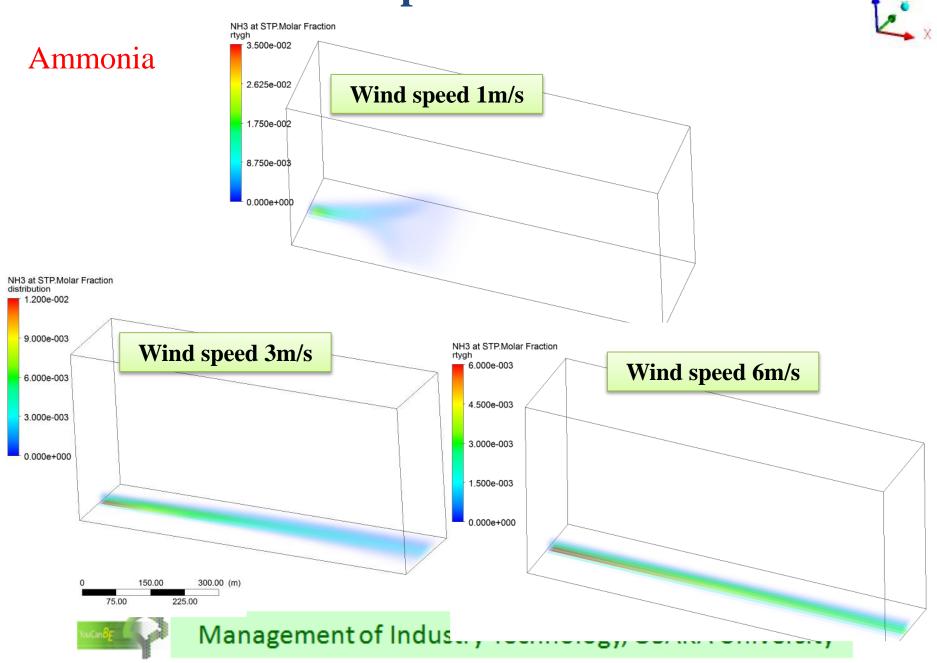


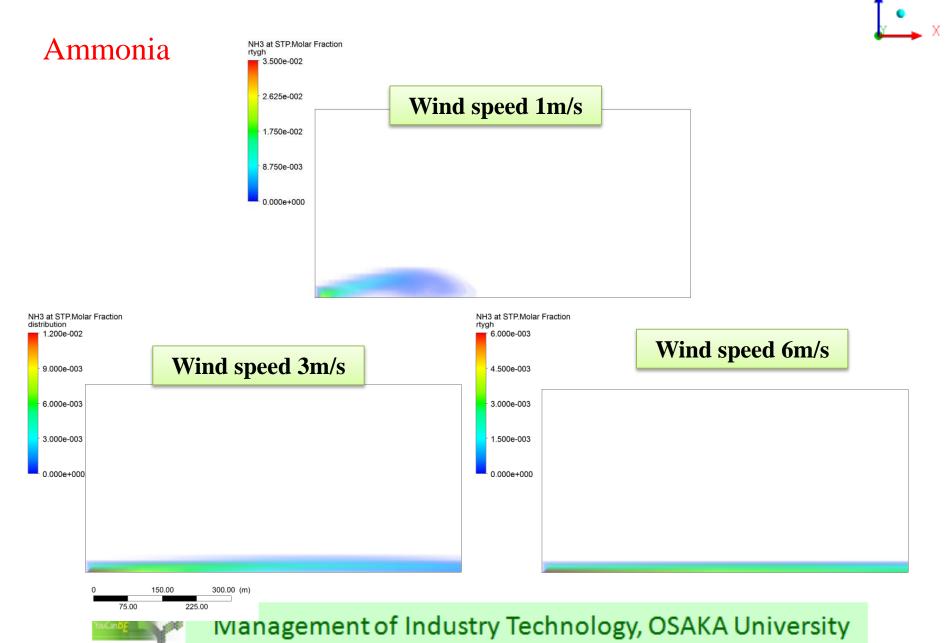
Air Temperature	25°C		
Anylysis time	Until 800s		

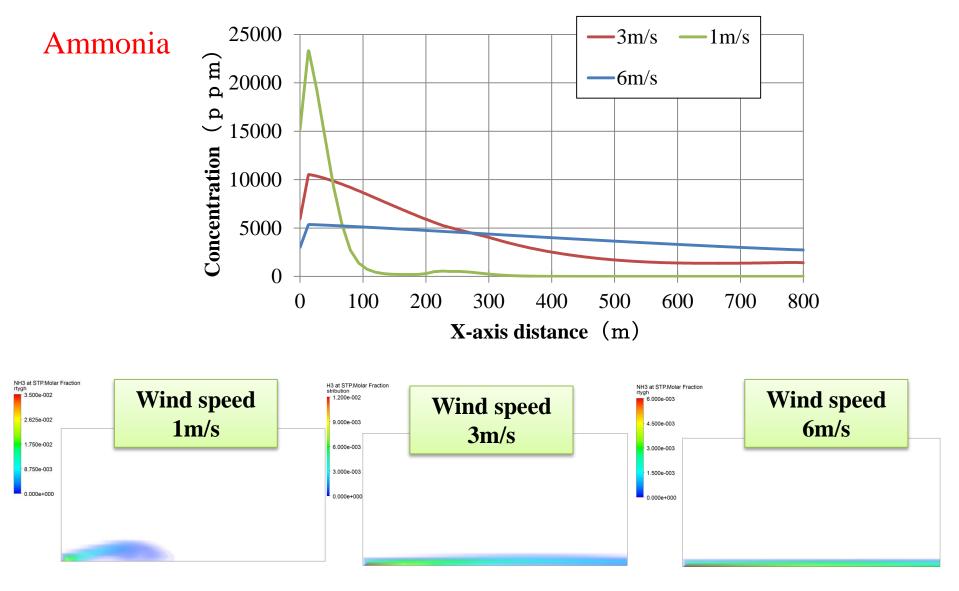
MESH



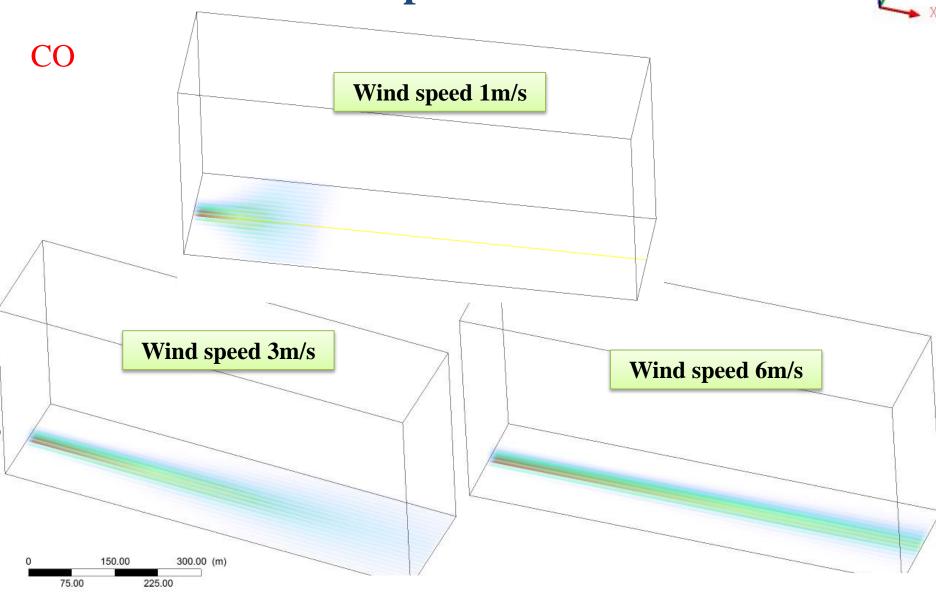
Statistics					
Nodes	30256				
Elements	27000				
Siz	ing				
Use Advanced Size Function	On: Curvature				
Relevance Center	Fine				
Smoothing	High				
Transition	Slow				
Span Angle Center	Fine				
Curvature Normal Angle	Default (18.0 °)				
Min Size	2.4e-002 m				
Max Face Size	Default (13.380 m)				
Max Size	Default (26.760 m)				
Growth Rate	Default (1.20)				
Minimum Edge Length	200.0 m				
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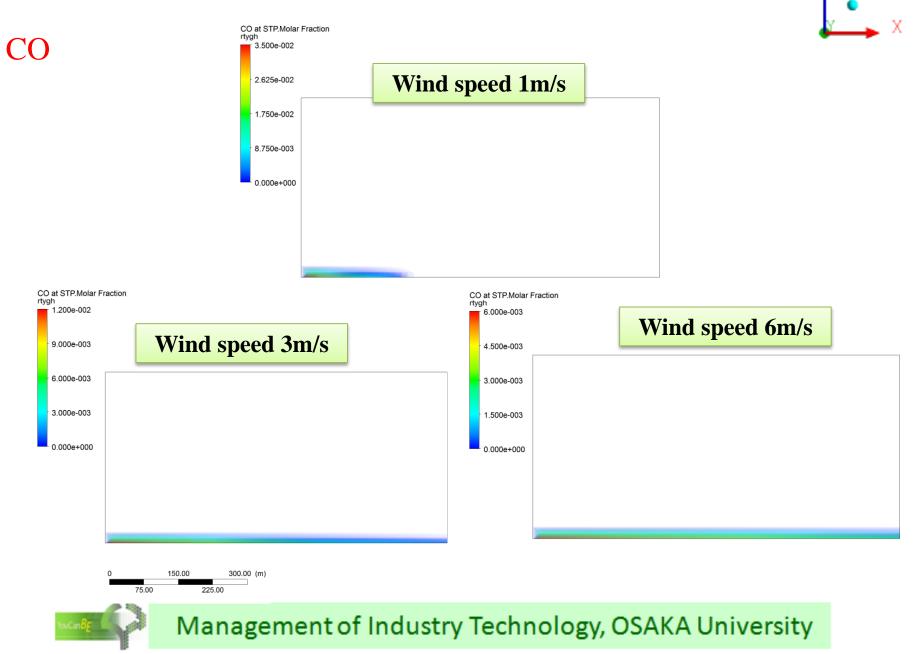


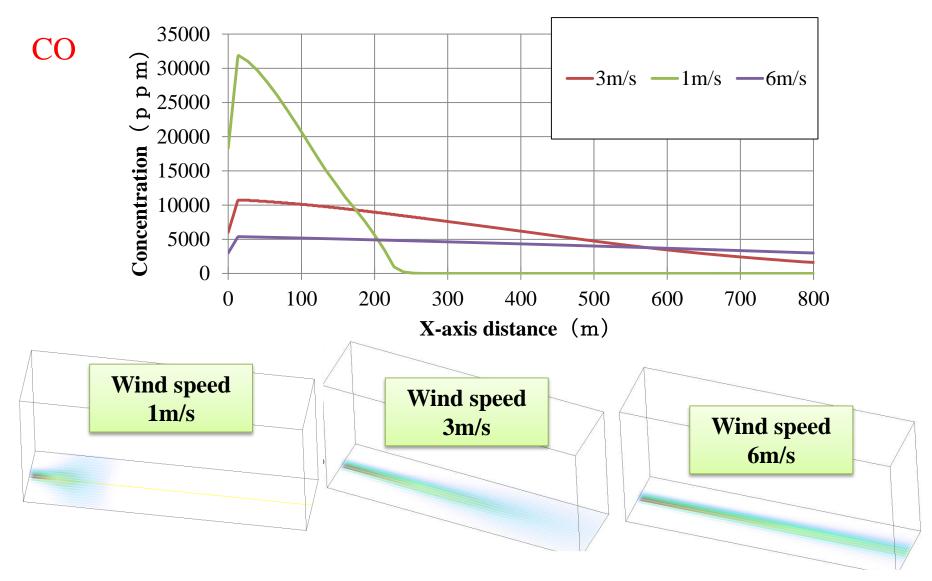




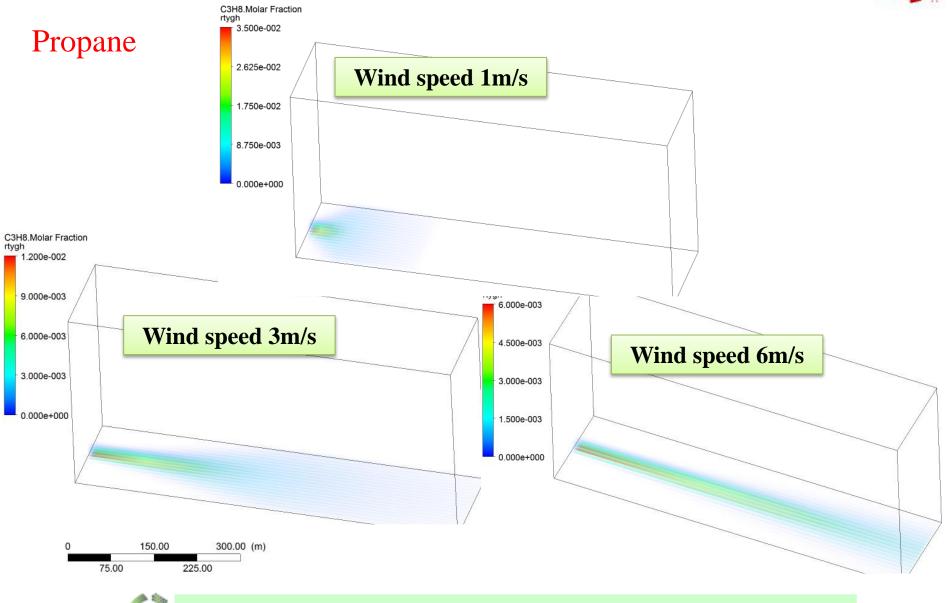




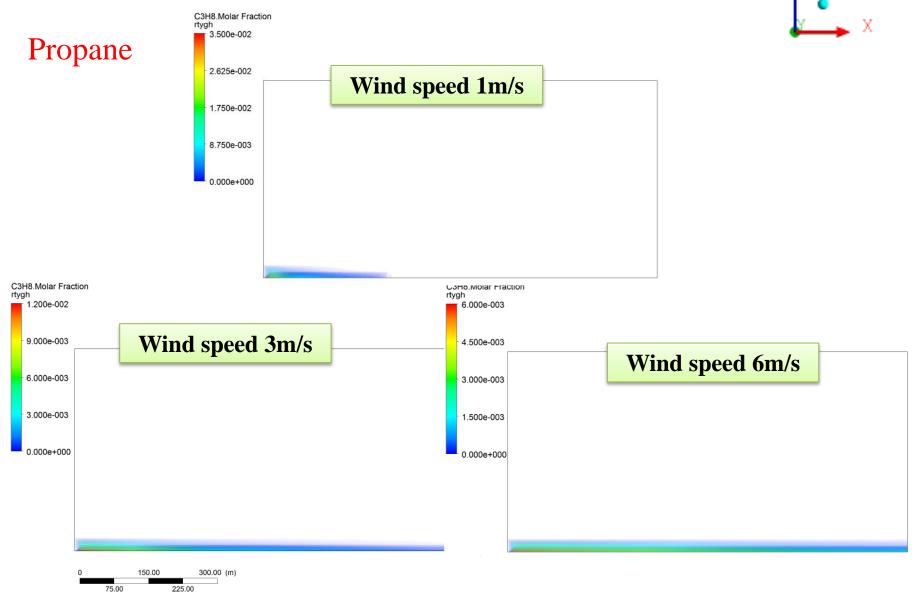




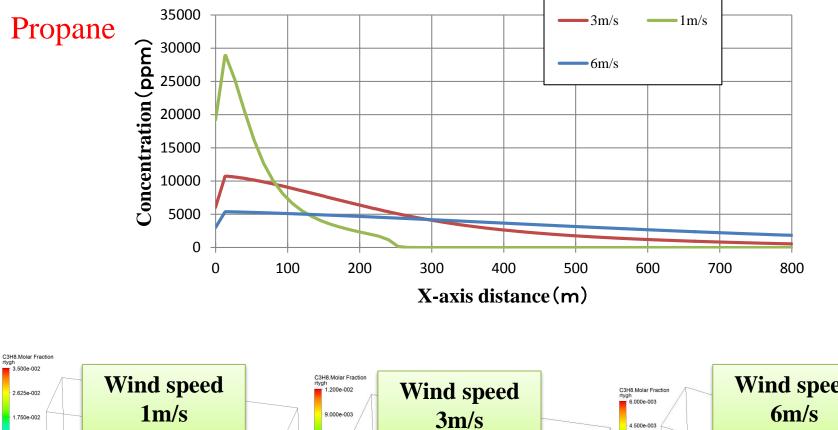


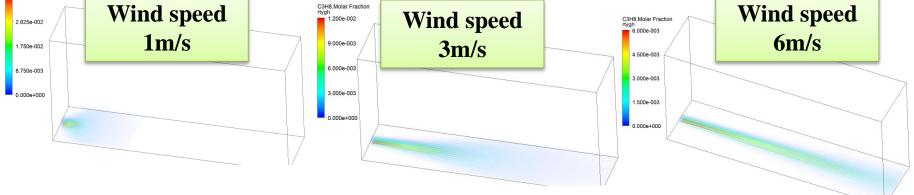


Yucande



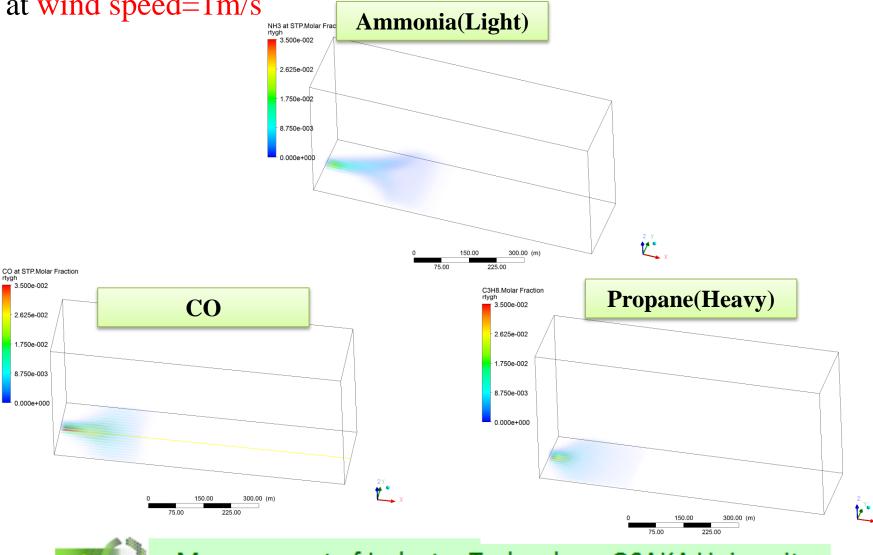
YuuCanBg





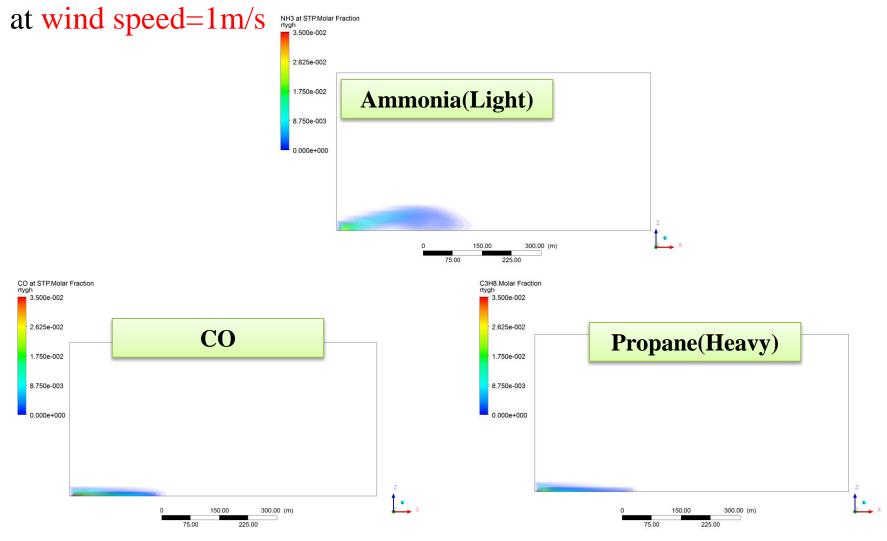


The distribution of gas concentration (side view) at wind speed=1m/s



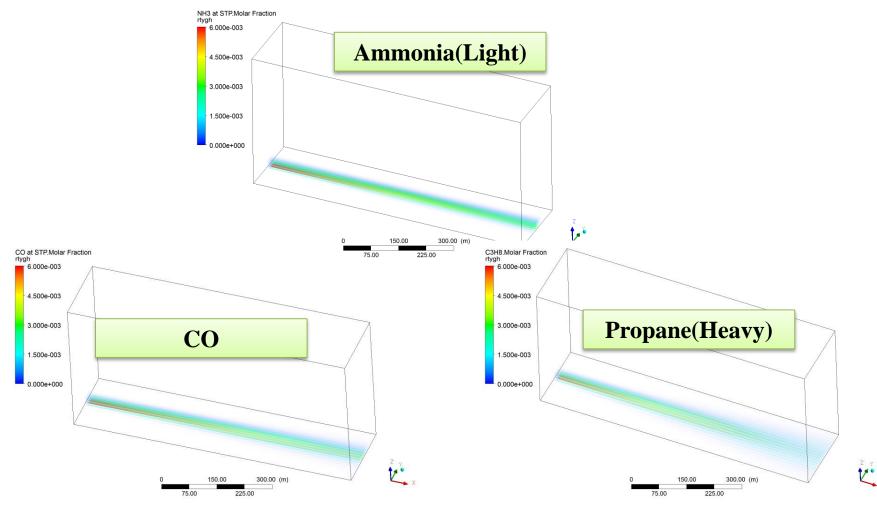
YuCinBg

The distribution of gas concentration (side view)





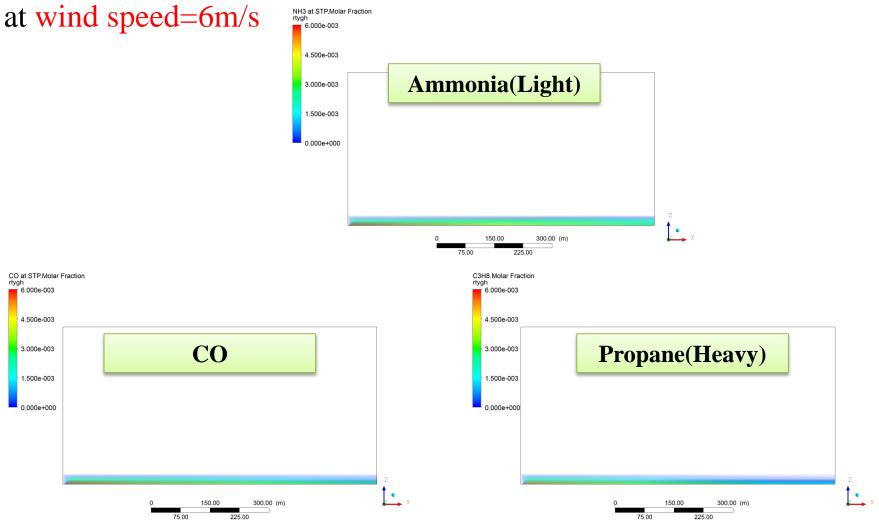
The distribution of gas concentration (side view) at wind speed=6m/s





The effects of different density

The distribution of gas concentration (side view)



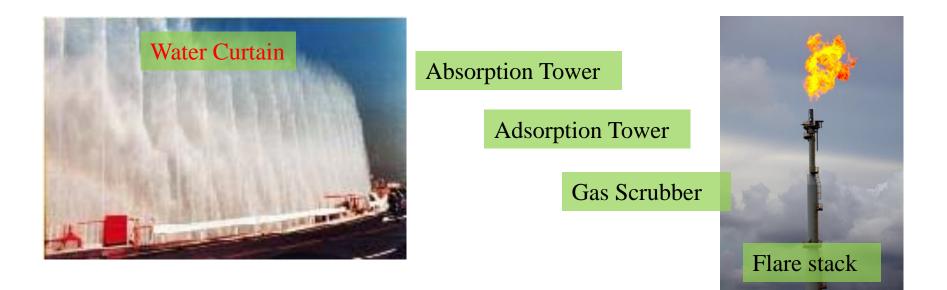


Gas diffusion analysis using numerical method

②The effects of mitigation equipment "Water spray curtain"



Examples of mitigation methods



Conventional diffusion analysis models are not able to consider the effect of mitigation measure



The effects of water curtain

Efficacy : <u>Reduction of the leeward side concentration</u>

Effects caused by the water curtain

Dispersion of diffused gas

-Change of gas dispersion caused by the water curtain

Absorption of diffused gas

-The effect that soluble substances dissolve into the water curtain



To estimate gas dispersion area correctly we have to consider these both effects.



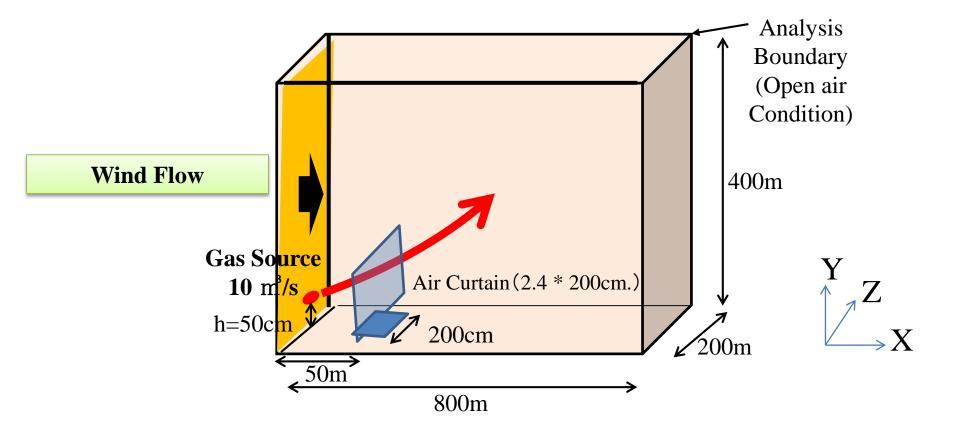
②Study gases and analysis conditions

Gas	Gas Density	Source Strength	Wind Speed	Absorption Level
Supposed Gas A	0.771kg/m³	10 m³/s	3m/s	No Curtain
Supposed Gas B	0.771kg/m³	10 m³/s	3m/s	None
Supposed Gas C	0.771kg/m³	10 m³/s	3m/s	High



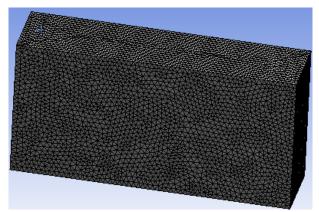
Analysis Model





Air Temperature	25°C
Nozzle water pressure	7 bar

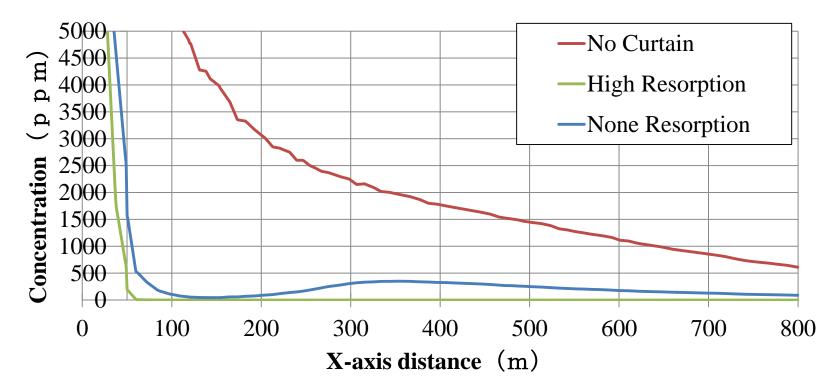
MESH



Statistics				
Nodes	25765			
Elements	133557			
Sizing				
Use Advanced Size Function	On: Curvature			
Relevance Center	Fine			
Smoothing	High			
Transition	Slow			
Span Angle Center	Fine			
Curvature Normal Angle	Default (18.0 °)			
Min Size	1.e-003 m			
Max Face Size	Default (13.380 m)			
Max Size	Default (26.760 m)			
Growth Rate	Default (1.20)			
Minimum Edge Length	4.e-002 m			
Management of Industry Technology, OSAKA University				

The efficiency of water curtain

Gas concentration at ground along x axis(leeward direction)

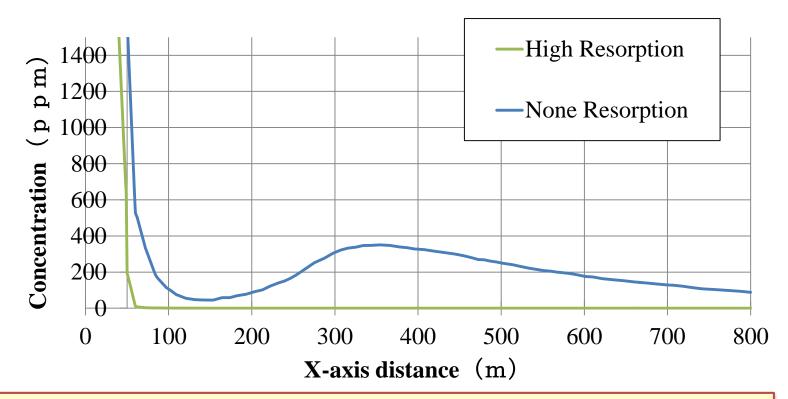


Gas concentration is largely reduced by water curtain



The efficiency of water curtain

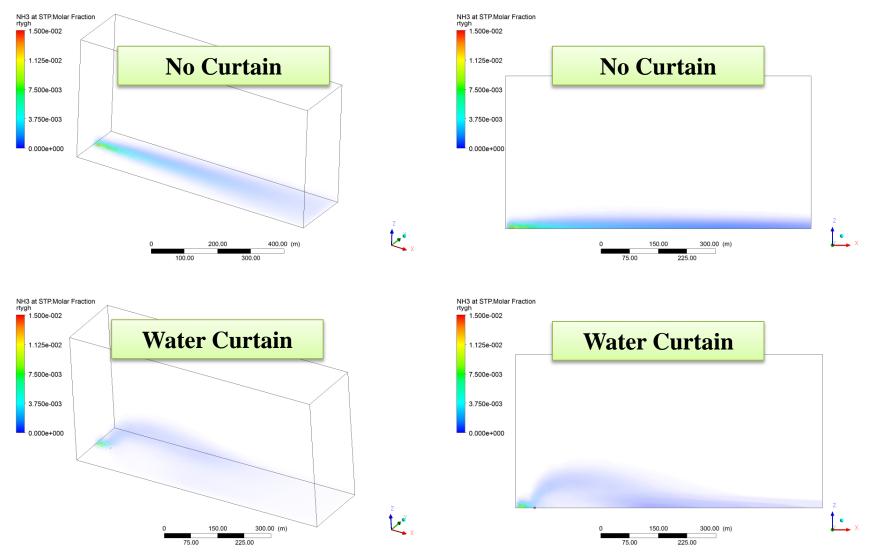
Gas concentration at ground along x axis(leeward direction)



The efficiency of water curtain can be increased by the absorption rate of liquid of curtain

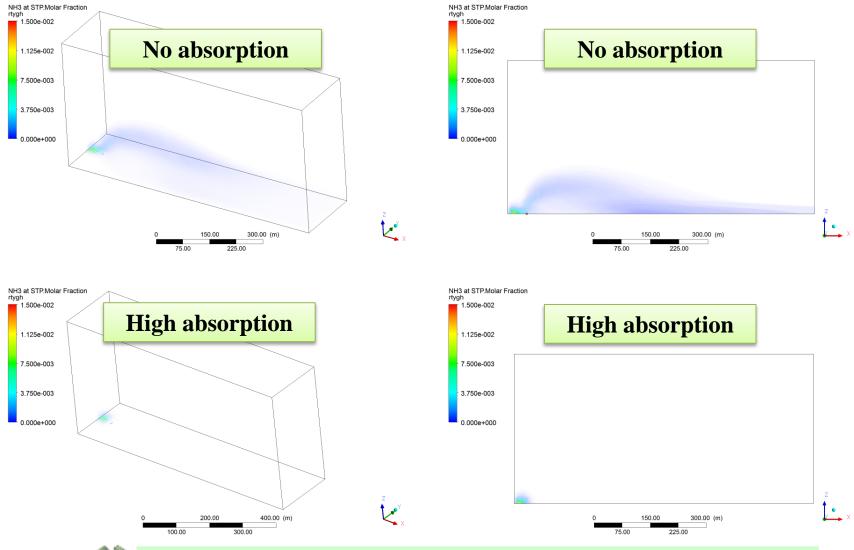


Results (Concentration Distribution Image)





Results (Concentration Distribution Image)





Summary



Overall Summary

- The effects of wind speed and density
 Problems of plume model
- -The important of mesh size for future study on CFD
- -The effects of wind speed and density on gas diffusion behaviors
- ^②The effects of "Water spray curtain"
- -Evaluated the effect of water spray curtain by simulated model
- -The important of absorption level



Thank you very much,

ありがとうございました。



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